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 $\text{HPO}_4^{2-} + \text{NH}_4\text{OH} = \text{H}_2\text{NPO}_4^- + \text{NH}_4^+$   
 $\text{H}_2\text{O} + 2\text{K}_2\text{SO}_4 + \text{BaCl}_2 \rightarrow \text{BaSO}_4 + 2\text{KCl}$   
 $\text{Ca(OH)}_2 + \text{NH}_4\text{Cl} \rightarrow \text{CaCl}_2 + 2\text{NH}_3 + 2\text{H}_2\text{O}$

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Chapter 7Chemical Reactions Section 7.1 Describing ReactionsMeasure Amounts Of A Substance Because Chemical Reactions Often Involve Large Numbers Of Small Particles. 10. Circle The Letter Of The Correct Answer. If One Carbon Atom Has An Atomic Mass Of 12.0 Amu And One Oxygen Atom Has An Atomic Mass Of 16.0 Amu, What Is The Molar Mass Of Carbon Dioxide? A.28.0 Amub.44.0 Amu C.28.0 Gd.44.0 G 11. 2th, 2024

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