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The Equations Below: 1) $1\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$ 2) $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$ 3) $2\text{NaCl} + 1\text{F}_2 \rightarrow 2\text{NaF} + 1\text{Cl}_2$ 4) $2\text{H}_2 + 1\text{O}_2 \rightarrow 2\text{H}_2\text{O}$ 5) $1\text{Pb}(\text{OH})_2 + 2\text{HCl} \rightarrow 2\text{H}_2\text{O} + 1\text{PbCl}_2$ 6) $2\text{AlBr}_3 + 3\text{K}_2\text{SO}_4 \rightarrow 6\text{KBr} + 1\text{Al}_2(\text{SO}_4)_3$ 7) $1\text{CH}_4 + 2\text{O}_2 \rightarrow 1\text{CO}_2 + 2\text{H}_2\text{O}$ Jan 19th, 2024

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