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Write Balanced Equations For The Following Redox Reactions: A. $2 \text{NaBr} + \text{Cl}_2 \rightarrow 2 \text{NaCl} + \text{Br}_2$ B. $\text{Fe}_2\text{O}_3 + 3 \text{CO} \rightarrow 2 \text{Fe} + 3 \text{CO}_2$ In Sour Solution C. $5 \text{CO} + \text{I}_2\text{O}_5 \rightarrow 5 \text{CO}_2 + \text{I}_2$ In The Basi Jul 1th, 2024.

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Oxidation With Nickel Peroxide. I. Oxidation Of AlcoholsNickel Peroxide In An Aqueous Alkaline Solution Afforded The Corresponding Carboxylic Acids, While Carbonyl Compounds Were Obtained On Oxidation Of Primary Or Secondary Alcohols By This Reagent In Organic Solvents. Although Such Metal Oxides As Manganese Di- Oxide, Selenium Dioxide, Lead Dioxide, Silver Oxide, Jun 4th, 2024

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