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 $2H^+ + OH^-$ What Would Alter The Composition Of The Equilibrium Mixture In Favour Of The V^{2+} Ions? A Adding An Acid B Adding A Reagent That Selectively Precipitates Apr 17th, 2024

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(c) (i) Write A Chemical Equation Representing The Lattice Energy Of AgBr. (ii) Use The Following Data To Calculate A Value For The Lattice Energy Of AgBr(s). First Ionisation Energy Of Silver = $+731\text{kJmol}^{-1}$ Electron Affinity Of Bromine = 325kJmol^{-1} Enthalpy Change Of Atomisation Of Silver = $+285\text{kJmol}^{-1}$ Enthalpy Change Of Atomisation Of Bromine = $+112\text{kJmol}^{-1}$... Mar 10th, 2024

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