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Equilibrium Mixture In Favour Of The V^{2+} Ions? A
Adding An Acid B Adding A Reagent That Selectively
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Equation Representing The Lattice Energy Of AgBr. (ii)
Use The Following Data To Calculate A Value For The
Lattice Energy Of AgBr(s). First Ionisation Energy Of
Silver = $+731 \text{ kJ mol}^{-1}$ Electron Affinity Of Bromine =
 325 kJ mol^{-1} Enthalpy Change Of Atomisation Of Silver
= $+285 \text{ kJ mol}^{-1}$ Enthalpy Change Of Atomisation Of
Bromine = $+112 \text{ kJ mol}^{-1}$... 8th, 2024.
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