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Mole Of HCl In 2.2 L Of Solution. $\text{HCl} \rightarrow \text{H}^+ + \text{Cl}^-$
Molarity Of HCl (which Is The H^+ Same As H_3O^+) =
 $6.2 \times 10^{-3} \text{ Mol} / 2.2 \text{ L} = 2.9 \times 10^{-3} \text{ M}$ [H 3 Mar 4th, 2024]

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