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WARM-UP (1) Convert 78 C To Kelvin 351 K (2) 3.45 Atm = \_\_\_\_\_ MmHg 2622 MmHg (3) The Coldest Temperature Possible Is (0 K) = Absolute Zero (4) What Is The Final Volume Of A 400.0 ML Gas Sample That Is Subjected To A Temperature Change From 22.0 °C To 30.0 °C And A Pressure Change From 760.0 Mm Hg To 360.0 Mm Hg? REMEMBER: Temp Must Be In Kelvin!! 1th, 2024

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