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Assigning Oxidation Numbers Directions: Assign Oxidation Numbers To All Of The Elements In Each Of The Compounds Or Ions Below. 1. HCl 11. H₂SO₃ 2. KNO₃ 12. H₂SO₄ 3. OH⁻ 13. BaO 2 4. Mg₃N₂ 14. KMnO₄ 5. KClO₃ 15. LiH 6. Al(NO₃)₃ 16. MnO₂ 7. S₈ 17. OF₂ 8. H₂O₂ 18. SO₃ 9. PbO 2 19. NH₃ 10. NaHSO₄ 20. Na

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AP CHEMISTRY NOTES 2-1 ASSIGNING OXIDATION ...

ASSIGNING OXIDATION NUMBERS RULES FOR ASSIGNING OXIDATION NUMBERS: 1. The Oxidation Number Of Any Free Element (including Diatomic Elements) Is Always 0. 2. The Sum Of The Oxidation Number Of All Of The Atoms In A Compound Is Always Equal To 0. 3. The Sum Of The Oxidation Numbers Of All Of Th

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ASSIGNING OXIDATION NUMBERS WORKSHEET

4. The Sum Of The Oxidation Numbers Of All The Elements In A Polyatomic Ion Equals The Charge On The Ion. Directions: In The Following Questions, Give The Oxidation Number Of The Indicated Atoms/ion. 1. N In N_2O_3 N^{3+} 2. S In H_2SO_4 S^{6+} 3. C 2CO 4. C In CO_2 $2+$ 5. Na In NaCl Na^+ 6. H I Mar 4th, 2024

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2. For Monoatomic Ions, The Oxidation Number Is The Same As The Charge. 3. All Metals In Group 1 Form $+1$ Ions And Group 2 Form $+2$ Ions When Found In A Compound. 4. When Found In A Compound Fluorine Has A -1 Oxidation State. 5. Hydrogen Has A $+1$ Oxidation State When Bonded To A Non Jul 4th, 2024

RULES FOR ASSIGNING OXIDATION NUMBERS

6. The Algebraic Sum Of The Oxidation Numbers Of Elements In A Compound Is Zero. CaF_2 $2\text{H}_2\text{SO}_4$ $+2$ +

$2(-1) = 0$ $2(+1) + S + 4(-2) = 0$ Find $S = +6$ For Ox.
No. Of S 7. The Algebraic Sum Of The Oxidation
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Assigning Oxidation#s Practice

Assigning Oxidation Numbers ! KEY Determine The
Oxidation Numbers To ALL Of The Elements In Each Of
The Compounds Or Ions Below: 1) HCl 8) H₂O 2 15)
LiH H = +1 H = +1 (each) Li = +1 Cl = -1 O = -1
(each) H = -1 ... Microsoft Word - Assigning
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Numbe R In IMA Name Chemical Formula (unformatted) Coll.

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Arsenohauchecornite $Ni_{18}Bi_3As_{16}$ 1 Arsenolamprite
As 1 Arsenolite As_2O_3 >2 Arsenopyrite $FeAsS$ >10
Arthurite $UFe^{+++}_2(AsO_4, PO_4, SO_4)_2(O, OH)_2 \cdot 4(H_2O)$ 1
Artinite $Mg_2(O_3)(OH)_2 \cdot 3(H_2O)$ >2 Apr 3th, 2024

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