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PH And POH - Ms. Mogck's Classroom - Ms. Mogck's Classroom

KEY Chemistry: PH And POH Calculations Part 1: Fill In The Missing Information In The Table Below. PH $[H_3O^+]$ POH $[$

OH1-] ACID Or BASE? 3.78 1.66 X 10⁻⁴ M 10.22 6.03 X 10⁻¹¹ M Acid 3.41 3.89 X 10⁻⁴ M 10.59 2.57 X 10⁻¹¹ M Acid 8.81
1.55 X 10⁻⁹ M 5.19 6.46 X 10⁻⁶ M Base 8.69 2.04 X 10⁻⁹ M 5.31 4.88 X 10⁻⁶ M Base 8.46 3.47 X 10⁻⁹ M 5.54 2.88 X 10⁻⁶ M
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