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 $\text{H}_2\text{O}(\text{l}) \rightleftharpoons \text{CH}_3\text{CH}_2\text{COO}^-(\text{aq}) + \text{H}_3\text{O}^+(\text{aq})$  Propanoic Acid,  
 $\text{CH}_3\text{CH}_2\text{COOH}$ , Is A Carboxylic Acid That Reacts With  
Water According To The Equation Above. At 25 C The  
PH Of A 50.0 ML Sample Of 0.20 M  $\text{CH}_3\text{CH}_2\text{COOH}$  Is  
2.79. (a) Identify A Brønsted-Lowry Conjugate Acid-  
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