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Zn(s) NaOH(aq) Y SECTION 11.3 REACTIONS IN AQUEOUS SOLUTION 1. Write The Net Ionic Equation For The Reaction Between Aqueous Barium Nitrate, Ba(NO<sub>3</sub>)<sub>2</sub>, And Sodium Sulfate, Na<sub>2</sub>SO<sub>4</sub> 2. Magnesium Reacts With HCl To Form Hydrogen And Magnesium Chloride. Write The Balanced Net Ionic Equation For This Reaction.

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Rearranged To Form Different Substances Products: Right Side Of The Reaction  
Reactants: Left Side Of The Reaction Chemical Equation: Represents With Symbols  
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Date Pd Unit 7 Lab - Describing Chemical ReactionsUnit 7 Lab - Describing Chemical  
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Of Basic Types Of Chemical Reactions. You Will Learn To Classify Basic Types Of  
Reactions And Write Balanced Equations, Including The Role Of Energy, To  
Effectively Communicate And Describe The Chemistry Of The Reactions. 3th,  
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