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The Purpose Of The Laboratory Is To Experimentally Determine How Temperature Influence The Solubility Equilibria Of A Salt With Low Solubility, $\text{Ca}(\text{OH})_2$. An Experimental Determination Of The Solubility Product Constant, K_{sp} , At Two Temperatures Will Be Employed To Determine H° , ΔS° , Δ And ΔG° For The Solubility Reaction. INTRODUCTION. Most Hydroxides (except Those Alkali Metals And ... 2th, 2024
Dependence Of The Solubility On Temperature
 M_i Is The Partial Molar Enthalpy And S_i Is The Partial Molar Entropy Of The i -th Component. Substituting This Into Eq. 3, We Get The Correlation $\ln x$

Saturated = H_0 Diss H M, solid R! 1 T + S_0 Diss S M, solid R (6) This Includes Partial Molar Quantities, Values That May Also Be Composi 3th, 2024.

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