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$\Delta S^\circ$ ,  $\Delta H^\circ$  And  $\Delta G^\circ$  For The Solubility Reaction. INTRODUCTION. Most Hydroxides (except Those Alkali Metals And ... Mar 1th, 2024) Dependence Of The Solubility On Temperature  $M_i$  Is The Partial Molar Enthalpy And  $S_i$  Is The Partial Molar Entropy Of The  $i$ -th Component. Substituting This Into Eq. 3, We Get The Correlation  $\ln x_{\text{saturated}} = \frac{H_0}{RT} + \frac{S_0}{R} - \frac{H_{\text{M,solid}}}{RT} - \frac{S_{\text{M,solid}}}{R}$  (6) This Includes Partial Molar Quantities, Values That May Also Be Computed Jul 1th, 2024.

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