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CH₂=CH₂

B - Polarity Like Alkenes, Alkene Are Relatively Nonpolar. They Are Insoluble In Water But Soluble In Nonpolar Solvents Such As Hexane, Gasoline, Halogenated Solvents, And Ethers. Alkene Tend To Be Slightly More Polar Than Alkanes, However, Because The More Weakly Held Electrons In The Pi

Bond Are More Polarizable. Jan 2th, 2024

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3

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