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### **15.3 Heterogeneous Aqueous Systems 15**

These Mixtures Are Classified As Solutions. In Contrast, Heterogeneous Mixtures Are Not Solutions. If You Shake A Piece Of Clay With Water, The Clay Breaks Into Fine Particles. The Water Becomes Cloudy Because 3th, 2024

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\_\_\_\_\_ 1. Which Of The Following Aqueous Mixtures Would Be A Buffer System? A. HCl, NaCl B. HNO<sub>3</sub>, NaNO<sub>3</sub> C. H<sub>3</sub>PO<sub>4</sub>, H<sub>2</sub>PO<sub>4</sub><sup>-</sup> D. H<sub>2</sub>SO<sub>4</sub>, CH<sub>3</sub>COOH E. NH<sub>3</sub>, NaOH 2. Which, If Any, Of The Following Aqueous Mixtures Would Be A Buffer System? 3th, 2024

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Typically, Intermolecular Surface Forces Are Interpreted In Terms Of Basic Interactions Such As Van Der Waals, Electron Donor-acceptor (hydrogen Bonding), And Electrochemical In-teractions.<sup>1</sup> Most Experimental Data On The Thermodynamics At The Interfaces Come From Studies Of Surface Or Interfacial Tension  
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