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CHAPTER 11 INTERMOLECULAR FORCES AND LIQUIDS AND ... CHAPTER 11: INTERMOLECULAR FORCES AND LIQUIDS AND SOLIDS 287 11.39 The Mass Of One Cube Of Edge 287 pm Can Be Found Easily From The Mass Of One Cube Of Edge 1.00 cm (7.87 g): $3 \times 10^{-3} \times 3 \times 10^{-3} \times 7.87 \text{ g} = 1 \text{ cm}^3 \times 10^{-3} \text{ M} (287 \text{ pm}) = 1.86 \times 10^{-3} \text{ g} / \text{unit cell}$ 1 cm 0.01 M 1 pm Jan 5th, 2024 Chapter 11: Intermolecular Forces And Liquids And Solids GCh11-4 Intermolecular Forces • Attractive Forces Between Molecules • Weaker Than Forces Within Molecules Between Atoms (intramolecular) • Stronger In Solids And Liquids Than In Gases • Responsible For Non-ideal Properties Of Gases Dipole Moment • A Vector Showing How An Electron Cloud Is Moved From One Part Of A Molecule To Another Part Feb 3th, 2024 Chapter 11 Intermolecular Forces, Liquids, And Solids Explaining Macroscopic Behavior: Viscosity • Viscosity Is The Resistance To Flow In Liquids • Viscosity Is Dependent On More Than Just Molecule Size - It Also Depends On The Kinds Of Attractive Forces Between Molecules • Viscosities Of Various Liquids At 20°C (in

Centipoise) Liquid Viscosity (cp) Type Of IM Attraction Feb 3th, 2024.

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