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1,000 1,100 1.200 1,300 Oxygen (Apr 2th, 2024.

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Moles Of All The Components Of The Mixture: $N = N_1 + N_2 + \dots$ It Follows That The
Sum Of All The Mole Fractions Of The Mixture Must Also Equal 1. $X_1 + X_2 + \dots = 1$
The Mass Of Component I In A Mixture Is The Product Of The Number Of Moles Of I
And Its Molecular Weight, M_i .The Mass Of The Mixture Is Therefore The Sum, $M =$
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