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Bonding: Covalent And Ionic (metallic Bonding Discussed Slightly In Ch 11) Examples: Cl_2 HCl NaCl Bond Classification Scheme Pure Covalent: No Difference In Electronegativity. Polar Covalent: Moderate Difference. Ionic: Large Difference. Really A Continuum (Fig. 8.7) Electronegati 1th, 2024

IONIC BONDING Name Ionic Bonding Occurs When A Metal ...

IONIC BONDING Name Ionic Bonding Occurs When A Metal Transfers One Or More Electrons To A Nonmetal In An Effort To Attain A Stable Octet Of Electrons. For Example, The Transfer Of An Electron From Sodium To Chlorine Can Be Depicted By A Lewis Dot Diagram, Calcium Would Need Two Chlorine Atoms To Get Rid Of Its Two Valence Electrons, 3th, 2024

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Tends To Lose One Electron From The Outer Level. 38.

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Or Group Of Atoms In A Chemical Species Is Known As
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Objectives For Today: Explain Properties Of Ionic And Metallic Compounds Using Diagrams. Bellwork: Get A White Board (one Per Table) and Answer The Questions Below: Draw The Lewis Dot Structure Of The Formation Of A Compound Made A Calcium And Phosphorus. What Is The Charge Of Each Ion Formed? W 1th, 2024

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Ionic And Metallic Bonding Multiple Choice Identify The
Choice That Best Completes The Statement Or
Answers The Question. ____ 1. What Are The Valence
Electrons Of An Atom? A. The Electrons That An Atom
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