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Matter And Changes Of States Of Matter Concept 3: Identify Types Of Changes In Matter (Melting, Deposition, Condensation, Solidification, Sublimation, And Evaporation) Concept 4: Use Jun 4th, 2024 Kinetic Molecular Theory Mean Kinetic Energy For A Mole Of Gas • For A Mole Of Ideal Gas, Mean Kinetic Energy,  $K$  Is Related To The Root Mean Squared Velocity,  $v_{rms}$ , By  $L$  The Mean Kinetic Energy Of A Sample Of Ideal Gas Is Directly Proportional To Absolute Temperature.  $L$  The Mean Apr 3th, 2024.

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 $L = \text{Liters}$ ,  $\text{Atm} = \text{Atmospheres}$ ,  $\text{Mol} = \text{Moles}$ ,  $\text{K} = \text{Kelvin}$ ,  $\text{J} = \text{Joules} = \text{Liters} \times \text{Kilopascals}$   
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