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13.1 Kinetic Theory And A Model For Gases The Word Kinetic

13.1 Kinetic Theory And A Model For Gases The Word Kinetic Refers To Motion. The Energy An Object Has Because Of Its Motion Is Called Kinetic Energy. According To The Kinetic Theory, All Matter Consists Of Tiny Particles Tha 1th, 2024

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CHAPTER 12 GASES AND KINETIC-MOLECULAR THEORY

3 Boyle's Law: The Volume-Pressure Relationship $V \propto 1/P$ Or $V = K (1/P)$ Or $PV = K$ $P_1V_1 = K_1$ For One Sample Of A Gas. $P_2V_2 = K_2$ For A Second Sample Of A Gas. $K_1 = K_2$ For The Same Sample Of A Gas At The Same T. \therefore Thus We Can Write Boyle's Law Mathematically As $P_1V_1 = P_2V_2$ 4th, 2024

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