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Mass To Mass Stoichiometry Problems Worksheet Answers SO_3 As A Conversion Factor, We Determine The Mass That This Number Of Moles Of SO_3 Has. The First Step Resembles The Exercises We Did In Section 6.4 "Mole-Mole Relationships In Chemical Reactions". As Usual, We Start With The Quantity We Were Given: $3.59 \text{ Mol } Fe_2O_3 \cdot \frac{3 \text{ Mol } SO_3}{1 \text{ Mol } Fe_2O_3} = 10.77 \text{ Mol } SO_3$ The Fe_2O_3 Units Cancel, Leaving $10.77 \text{ Mol } SO_3$... 1th, 2024

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