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The Equation: $\text{N}_2 + 3\text{H}_2 \rightleftharpoons 2\text{NH}_3$ (1) $H_0 = 92 \text{ KJ Mol}^{-1}$

$S_0 = 192 \text{ J Mol}^{-1} \text{ K}^{-1}$ 2th, 2024

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5 1.6 Q.1 2th, 2024

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