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Directions: Solve Each Of The Following Problems.
Show Your Work, Including Proper Units, To Earn Full
Credit. 1. Silver And Nitric Acid React According To The
Following Balanced Equation: $3 \text{Ag(s)} + 4 \text{HNO}_3 \text{(aq)} \rightarrow 3 \text{AgNO}_3 \text{(aq)} + 2 \text{H}_2\text{O(l)} + \text{NO} \dots$ Feb 1th, 2024.
Chemistry Supplemental Stoichiometry Problems

Answer Key Chemistry Supplemental Stoichiometry Problems Answer Key
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Chemistry 11 Stoichiometry Review Worksheet Name: KEY
 Chemistry 11 Stoichiometry Review Worksheet Name: ____ KEY ____ 33. $\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$
 Actual yield = 1.45g Theoretical yield = 5.95g $\text{KClO}_4 \times 1\text{mol} = 122.55\text{g} \times 2\text{O}_2 = 32.00\text{g} \times 1\text{mol} = 3.11\text{g O}_2$
 $\% \text{Yield} = \frac{1.45\text{g}}{3.11\text{g}} \times 100\% = 46.6\%$ 34. $4\text{FeCO}_3 + \text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$
 Percent purity = 62.8%? g pure FeCO_3 Jul 2th, 2024
 Practice Work 53 Stoichiometry-04 Mixed

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Chapter 4 Aqueous Reactions And Solution
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