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33. $\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$ Actual yield = 1.45g Theoretical yield = 5.95g $\text{KClO}_4 \times 1 \text{mol} \times 122.55 \text{g} \times 2 \times 2 \times 1 \text{KClO}_4 \times 32.00 \text{g} \times 1 \text{mol} = 3.11 \text{gO}_2$
% Yield = $\frac{1.45 \text{g}}{3.11 \text{g}} \times 100\% = 46.6\%$
34. $4\text{FeCO}_3 + \text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$ Percent purity = 62.8%
pure FeCO_3 Jul 2th, 2024Practice Work 53 Stoichiometry-04 Mixed Stoichiometry ... Practice Work 53 - Stoichiometry-04 Mixed Stoichiometry Problems General Information You Will Need A Periodic Table, Your Stoichiometry Notes, And Appendix 12 For This Assignment. Sorry About The Lack Of Format. I'm In A Time Crunch. 123.88 G/mol 70.90 G/mol 137.32 ... May 4th, 2024Chapter 12 Stoichiometry Review Worksheet Answer KeyAcces PDF Chapter 12 Stoichiometry Review Worksheet Answer Key Chapter 12 Stoichiometry Review Worksheet Answer Key Yeah, Reviewing A Books Chapter 12 Stoichiometry Review Worksheet Answer Key Could Accumulate Your Close Contacts Listings. This Is Just One Of The Solutions For You To

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