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Mole Is The Number Of Atoms Which Has A Mass In Grams Equal To The Numerical Value Of The Mass Of The Atom In Atomic Mass Units (amu). So The Mass Of An Average Carbon Atom Is 12 Amu And The Mass Of A Mole Of Carbon Atoms Is 12 G. Pogil Answer Key

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Mole Ratios Practice Worksheet Answer Key What Would Be The Weight Of This Maximum Yield? To Answer This, We Need To Convert Our Mole Value For  $\text{H}_2\text{S}$  To A Mass Value. Remember That  $M = nM_w$  We

Can Specify  $M_w = (2 \times 1,008) + 32.06 = 34.08 \text{ g/mol}$   
Proto  $M = 3.5 \times 34.08 = 119.3 \text{ grams}$  H<sub>2</sub>S could be produced.  
III. If you would like to produce 3.00 kg H<sub>2</sub>S, 2024  
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So the mass of an average carbon atom is 12 amu  
and the mass of a mole of carbon atoms is 12 g.  
The activity began with a set of data that showed  
when there are equal numbers of quail eggs and.  
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