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Reaction Is Given Below: $2\text{NO}(\text{g}) + \text{Br}_2(\text{g}) \rightleftharpoons 2\text{NOBr} \dots$
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And Water Does Not Change With Time. (ii) The
Intensity Of Red Colour Increases When Oxalic Acid Is
Added To A Solution Containing Iron (III) Nitrate And
Potassium Thiocyanate. (iii) On Addition Of The
Catalyst, The Equilibrium Constant Value Is Not
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Methane, Graphite And Dihydrogen At 298 K Are,

-890.3 KJ Mol⁻¹ -393.5 KJ Mol⁻¹, And -285.8 KJ Mol⁻¹

Respectively. Enthalpy Of Formation Of CH₃th,

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Thermodynamics Is Not Concerned . About. (a) Energy

Changes Involved In A Chemical Reaction. (b) The

Extent To Which A Chemical Reaction . Proceeds. (C)

The Rate At Which A Reaction Proceeds. (d) The

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(i) Used To Determine Heat Changes (ii) Whose Value

Is Independent Of Path (iii) Used To Determine

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