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Apr 5th, 2024 Introduction To Nuclear Physics And Nuclear Decay The Binding Energy Per Nucleon Dictates Results When Nuclides Break Apart (fission) Or Fuse Together (fusion) (keep In Mind That ... (two Protons And Two Neutrons) ! " = $2\text{ }^4\text{He} + 2\text{ }^0\text{e}^- + \bar{\nu}_e$ Or An Anti-neutrino ($\bar{\nu}_e$) Neutrinos Have No Charge, Spin $1/2$, And Mass $\sim 0.1 - 1\text{ eV}$ (?) The Beta Particle

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Rates $N(t) = N_0 e^{-kt}$ Where $N(t)$ Is The Number Of Atoms
Remaining At Time t , $N_0 = N(0)$ Is The Initial Number
Of Atoms, And $k = \ln(2)/T_{1/2}$ Is A Constant Deter-
mined By The Nuclear Half-life $T_{1/2}$. However, Recent
Work Has Uncovered Evidence Of An Unexpected
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Nuclear Equations: 1. Alpha Decay Of Radium-226, The

Most Abundant Isotope Of Radium $^{226}\text{Ra} + \text{He } 88$ 2.

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Nuclear Chemistry Alpha, Beta, And Gamma Decay Practice
Balancing Decay Reactions: Fill In The Blank In Each Of The Following Decay Reactions With The Correct Decay Particle Or Decayed Nucleus That Will Balance The Reaction, Also State Whether It Is Alpha Or Beta Decay
15. ${}^{184}_{74}\text{W} \rightarrow {}^{180}_{72}\text{Yb} + \text{_____}$ Is An Example Of _____ Decay.
16. ${}^{35}_{16}\text{S} \rightarrow \text{_____} + {}^{35}_{17}\text{Cl}$
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