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Development Of The Periodic Table In The Late 1700s, French Scientist Antoine Lavoisier (1743-1794) Compiled A List Of All Elements That Were Known At The Time. The List, Shown In Table 6.1, Contained 33 Elements Organized In Four Categories. Many May 4th, 2024.

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THE PERIODIC TABLE & PERIODIC LAWE. Modern Periodic Table 1. Modern Periodic Law: Similar Properties Of The Elements Occur Periodically When Arranged By Their Atomic Numbers (not Masses) 2. The Arrangement Of The Periodic Table Is Now Based On How Electrons Fill Various Energy Levels 3. Organization: A. Period: Horizontal Row Of Elements By Increasing Atomic Number; Jan 4th, 2024
PERIODIC TABLE AND PERIODIC PROPERTIES(g) The Metals Present In The Period 3. Ans. Sodium, Magnesium And Aluminium. (h) The Non-metals Present In Period 2 Ans. Carbon, Nitrogen, Oxygen And Fluorine. (i) The Group Of Elements Having Zero Valency. Ans. Group 18 Or Zero Group. (j) A Non-metal In Period May 3th, 2024
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