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Following Equation:  $\text{Fe}_2\text{O}_3 + 3 \text{H}_2 \rightarrow 2 \text{Fe} + 3 \text{H}_2\text{O}$ .

Calculate How Many Grams Of Iron Can Be Made From

16.5 Grams Of  $\text{Fe}_2\text{O}_3$  By The Following Equation. 1th,

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Simulation The Skater Had More Potential Energy At

The Bottom And Had More Kinetic Energy At The Top.

The Relationship Between Kinetic And Potential Energy

Is That When One Goes Up, The Other Comes Down.

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