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5.6 5.8 Kinetic Molecular Theory (KMT) Gas Particles Can Be Considered Point Particles Colliding With The Walls Of An Enclosure Randomly. The Higher The Temperature, The Faster Molecules Move, The More Kinetic Energy They Have. Kinetic Theory Gas Model For KM Mar 3th, 2024

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13.1 Kinetic Theory And A Model For Gases The Word Kinetic

13.1 Kinetic Theory And A Model For Gases The Word Kinetic Refers To Motion. The Energy An Object Has Because Of Its Motion Is Called Kinetic Energy. According To The Kinetic Theory, All Matter Consists Of Tiny Particles Tha Mar 3th, 2024

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1. Describe How Gases, Liquids, And Solids Compare Using The Following Table. Solids Liquids Gases Volume (definite Or Indefinite) Molecular Motion (high, Med, Low) Distance Between Molecules (g 2. Read Page 420: What Is Kinetic Energy? Describe Kinetic Molecular ... Feb 3th, 2024

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CHAPTER 12 GASES AND KINETIC-MOLECULAR THEORY

3 Boyle's Law: The Volume-Pressure Relationship { $V \propto 1/P$ Or {V = K (1/P) Or PV = K{P1V 1 = K 1 For One Sample Of A Gas. {P2V 2 = K 2 For A Second Sample Of A Gas. {k1 = K 2 For The Same Sample Of A Gas At The Same T. {Thus We Can Write Boyle's Law Mathematically As P 1V 1 = P 2V 2 May 2th, 2024

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Absolute Scale. D) There Is No Difference Whatsoever Between Them. 20. What Is The Difference Between The Mar 2th, 2024

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Converting Kelvin To Celsius - 273 Convert 393K To C° - 273 = 20°C Converting Fahrenheit To Celsius (not Used Much) Use C = 5/9(F - 32) Example: Convert 80°F To °C C = 5/9(F - 32) = 26.7°C. Physics 20: Heat Teacher Notes Page 8 Of 24 Practice Problems K = C + 273 C = K - 273 1. Convert Th Feb 3th, 2024

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