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A) How Many Moles Of Potassium Dichromate ( $\text{K}_2\text{Cr}_2\text{O}_7$ ) Are Required ... = 407.9 G AgBr This Is The Theoretical Yield Yields 22  
B) Calculate The Percent Yield If 375.0 G Of Silver Bromide Was Obtained From The Reaction  
Theoretical Yield = 407.9 G AgBr Percent Yield =  $100 \times \frac{\text{Actual Yield}}{\text{Theoretical Yield}}$   
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