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Steps To Predicting The Products Of Chemical ReactionsChemical Reactions Chemical Equations Chemical Equations • Balancing Equations • Predicting Products From Reactants • Synthesis • Decomposition • Single Replacement • Double Replacement • Combustion Packard, Jacobs, Marshall, Chemistry Pearson AGS Globe, Page 175 Mar 9th, 2024Answers For Predicting Products Of Chemical ReactionsAnswers For Predicting Products Of Chemical Reactions For All Combustion Reactions Of Hydrocarbons Or Hydrogen-carbon-oxygen Molecules, The Products Will Always Be CO 2 And H 2 O (assuming A Complete Reaction). A. Combustion: C 6 H 12 + 90 2 6CO 2 + 6H 2 O B. Combustion: 2C 4 H 6 + 110 2 8CO 2 + 6H 2 O C. Combustion: C 6 H 10 O 3 Feb 7th, 2024Predicting Products Of Chemical Reactions - Practice ProblemsName: Suggested Answers Date: _____ Predicting Products Of Chemical Reactions - Practice Problems Directions: Predict The Products For, And Then Balance Each Of The Following Chemical Reactions: 1. Sil4 + Mg (single Replacement) Sil4 + 2Mg 2Mgl2 + Si 2. 2Al + 3l2 (synthesis) 2Al + 3l2 2All3 3. CuC Feb 2th, 2024.

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