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Which Row Shows The Empirical Formula And Molecular Formula For This Alcohol? 8 The Equation Represents The Reaction Between Solid Magnesium Oxide And Dilute Hydrochloric Acid To Form Magnesium Chloride And Water. $\text{MgO} + 2\text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2\text{O}$ Which Row Shows The State Symbols For Hydrochl 1th, 2024

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AP* Stoichiometry Free Response Questions KEY Page 3 1991 2) A) Three Points $7.2 \text{ g H}_2\text{O} \div 18.0 \text{ g/mol} = 0.40 \text{ Mol H}_2\text{O}$
 $0.40 \text{ Mol H}_2\text{O} \times (2 \text{ Mol H} / 1 \text{ Mol H}_2\text{O}) = 0.80 \text{ Mol H}$ $7.2 \text{ L CO}_2 \div 22.4 \text{ L/mol} = 0.32 \text{ Mol CO}_2$ $0.32 \text{ Mol CO}_2 \times (1 \text{ Mol C} / 1 \text{ Mol CO}_2) = 0.32 \text{ Mol C}$ OR $N = PV \div RT = [(1 \text{ Atm}) (7.2 \text{ L})] \div [(0.0821 \text{ L Atm Mol}^{-1} \text{ K}^{-1}) (273 \text{ K})] = 0.32 \text{ Mol CO}_2$ 2th, 2024

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