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STOICHIOMETRY ...42 Chapter 2 Environmental Chemistry Section 2.2 Stoichiometry  
43 We Already Know That There Are 44 Grams Per Mole Of CO" So We Do Not Need  
To Recalculate That. Two Moles Of Butane ( $2 \text{ Mol} \times 58 \text{ G/mol} = 116 \text{ G}$ ) Yields 8  
Moles Of Carbon Dioxide ( $8 \text{ Mol} \times 44 \text{ G/mol} = 352 \text{ G CO}_2$ ) , So We Can Set Up The  
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Of Bananas, Can Be Produced Commercially. Interestingly, Bees Release About  $1\mu\text{g}$   
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