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$C_2H_5OH(l) + 3 O_2(g) \rightarrow 2 CO_2(g) + 3 H_2O(g)$ (31) A 100.0 mL Sample Of 0.300 M NaOH Is Mixed With A 100.0 mL Sample Of 0.305 M HNO_3 (31) In A Coffee Cup Calorimeter. If Both Solutions Were Initially At 35.0 °C And The Temperature Of The Resulting Solution Was Recorded As 37.0 °C, Determine The ΔH_{rxn} (in Units Of KJ/mol). 3th, 2024

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11.3: Heat In Changes Of State Enthalpy (H): - The Amount Of Internal Energy At A Specific Pressure And Volume. Energy Involved In Physical Change (Temperature Or Phase Change): Heating Curve: - A Graph Of Temperature Versus Time As A Substance Is Heated From A Solid Phase To A Gaseous Phase. 3th, 2024

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