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2) Aniline And Acetanilide Stand In A One-to-one Molar Relationship, So 0.0322 Mol Of Aniline Produces 0.0322 Mol Of Acetanilide. The Amount Of Acetic Anhydride Does Not Play A Role, Since It Is In Excess. Then, $n(\text{N}) = 0.0322$.aniline $n(\text{Acetanilide}) = 0.0322$ Mol We Find The Molar Mass Of Acetanilide: $(\text{N}) = (12 + 8 + 1 + 9 + 14 + 1 + 16 + 1 + 135) = 266$. 89 M Aceta ... Jun 1th, 2024

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